QUANTITATIVE ANALYSIS OF TOTAL PHENOLICS, FLAVONOIDS AND ANTIOXIDANT ACTIVITY OF OLIVE FRUITS (*OLEA FERRUGINEA*) BASED ON GEOGRAPHICAL REGION AND HARVESTING TIME IN ZHOB DISTRICT, PAKISTAN

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Abstract. Present study aims to explore fluctuating levels of total phenolic content (TPC), total flavonoid content (TFC) and antioxidant activity (AA) in fruits of *Olea ferruginea* Royle with respect to changing altitude, slope direction and harvesting time in District Zhob, Pakistan. Sampling was performed from three altitudes with approximate difference of 183 meter (600 feet) between each point at north facing and south facing slopes. Fruits were collected from twenty trees around each point, thoroughly washed with running water, shade dried, grinded and extracted against four solvents *i.e.*, water, acetone, 80% ethanol and 80% methanol. Extracts were further analyzed for TPC, TFC and AA (as DPPH radical scavenging activity). Solvent efficiency appears as acetone >80% methanol > 80% ethanol > water for maximum extraction of TPC, TFC and AA. Analysis of variance (ANOVA) and least significance difference (LSD) tests depicted that olives grown on south facing slopes have significantly higher TPC values as compared to north facing slopes whereas a reverse pattern was observed for TFC. Minimum TPC, TFC and AA was recorded in olives collected from middle altitude that increased with altitude. TPC, TFC and AA showed an increase when fruit fully matured while a drop was observed when green fruit turned purple.

Keywords: altitude, slope, ethanol, methanol, acetone

Introduction

Zhob district, Balochistan is well recognized for its naturally occurring population of wild olive trees. It is located in an agro-ecological zone and occupies 126,719 hectares of agricultural land. It is significant for agriculture and well known for its naturally occurring wild olive forest population. Olive and its various components are valued for their

functional food components and bioactive nutrients that promote health (Ghanbari et al., 2012).

The fruits of *Olea ferruginea* are edible, pickled and utilized as appetizers, antidiabetics and has emmenagogue substances. Olives fruits oil is efficient in oleic acid, utilized for curing scabies, typhoid, eye burning, jaundice, biliousness, toothache and teeth caries (Zabihullah et al., 2006; Ahmad, 2007). Fresh fruit of *Olea ferruginea* Royle in summer season are collected, dried and recommended to diabetics in winter season for reducing blood glucose level in District Attok, Pakistan (Ahmad et al., 2009).

Fruit of *Olea ferruginea* is commercially overlooked and underutilized in Pakistan due to its smaller size apparently. Cultivated varieties of Olive hold important place commercially especially in pharmaceutical and food industry. Phytovhemical composition varies in accordance with light availability in slopes, and altitudes (Måren et al., 2015) and on ripening stages. The microclimate has strong relationships with the direction of slope of the area as it influences the topography and the amount of solar radiation received by a specific slope (Sariyildiz et al., 2005). North facing slopes receive minimum amount of radiation and are therefore cool, moist and subject to slow changes in seasonal and daily microclimate. South facing slopes, however, receive maximum solar radiation so they are typically hot and dry and subject to rapid changes in seasonal and diurnal microclimate. Studies have proven that at different stages of development of the plant, the quality and the quantity of total phenols change (Amiot et al., 1986).

Cultivated varieties of Olive hold important place commercially especially in pharmaceutical and food industry. Few studies are available on phenolic contents and antioxidant activities of fruit of *Olea ferruginea* and no work is conducted on fluctuation in phenolics and antioxidant activities of olive fruit with changing slopes, altitudes and ripening stages. Present study deals with these issues. It can contribute to the identification of the best harvesting time, elevation and slope direction that can help to avail maximum phytochemical ingredients of fruit. The main objective of this study is to unveil the influence of geographical positioning and maturation stage on the level of TPC, TFC and AA of *Olea ferruginea* fruit in an area with forest ecosystem and huge climatic and geological variations.

Materials and methods

Geographical representation of the area with the marked slopes and altitudes were taken. The area was first studied and discussed in the Department of Excellence in Mineralogy, University of Balochistan with the help of toposheets of district Zhob. Brunton® Compass and Global Positioning System (GPS) were used to select the sites for sample collection. The area map was prepared in Geological survey of Pakistan Quetta (*Fig. 1*). *Olea ferruginea* was identified with Flora of Pakistan (Grohmann, 1974) and was confirmed by Dr. Rasool Bakhsh Tareen (qualified plant Taxonomist, University of Balochistan Quetta).

On the basis of contrasting features of north facing and south facing slopes, these two type of slopes were chosen for sampling at different altitudes, at different growth stages of fruit.

The microclimate of area was evaluated with the help of monthly average data of rain fall, relative humidity and temperature for the year 2017 was obtained from Regional Metrological Centre, Pakistan Metrological Department. The 10th January was the coldest day of the year with -5°C while the 15th June was the hottest day of the year with 41°C. Maximum rainfall was observed on the 14th November (28 mm) (*Fig. 2*).

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Figure 1. Sampling area map prepared on Terra incognita and Geographic Information System(GIS)



Figure 2. Metrological data of average monthly rainfall and average monthly temperature for the year 2017 obtained from Regional Metrological Centre, Metrological Department Pakistan

Chemicals and reagents

The following chemicals and reagents were used: Methanol (for HPLC \geq 99.9% Sigma-Aldrich), Acetone (Sigma-Aldrich), Ethanol (Sigma-Aldrich), Distilled water,

Folin-Ciocalteu's phenol reagent (purchased from BDH), Na₂CO₃, NaNO₂, AlCl₃, NaOH, Gallic acid, Catechin, diphenyl-picrylhydrazyl (DPPH), Butylated hydroxytoluene (BHT).

Fruit sampling

Local community call Zhob olives "Shnaney", while in urdu olives are known as Zaitoon. Stratified randomized sampling was conducted thrice in a year during crop year 2017-18. North facing and south facing slopes with dense population of test species were further divided in three strata based on elevation difference of approximately 600 feet in between each point. Around each point sampling was performed in a fashion to maintain height and slope direction with the help of Global Positioning System and Brunton compass (*Table 1*). First fruit sampling was conducted on 26th June when green olives appeared on trees, second time purple, ripe olives were picked on 26th August while last sampling was conducted on 26th October when fruit color turned mature blackish purple. From each sampling site fruits were picked randomly from twenty trees and mixed in one zip lock bag to prepare a representative batch of specific site.

| Table | 1. | Coordinates, | elevation | and | slope | direction | of | sampling | sites | recorded | with |
|--------|----|--------------|-----------|--------|---------|-----------|----|----------|-------|----------|------|
| Brunto | n® | Compass and | Global Po | sitior | iing Sy | stem (GPS |) | | | | |

| S.No. | Sampling site code | Slope direction | Dip | Latitudes | Longitudes | Elevation (Meters) |
|-------|-----------------------|--------------------|-----------------|---------------|----------------|-----------------------|
| 1 | TOP 1 | North facing | $45^{\circ} N$ | 31°16.637 N | 069°32.033 E | 1941m |
| 2 | MID 1 | North facing | 10°N | 31°29 57.98 N | 069°22 29.76 E | 1758 m |
| 3 | BOTTOM 1 | North facing | 65°N | 31°30 22.15 N | 069°20 58.66 E | 1576 m |
| 4 | TOP 2 | South facing | $20^{\circ}S$ | 31°17.0690 N | 069°32.836 E | 1944 m |
| 5 | MID 2 | South facing | 25°S | 31°17.42 N | 069°34 43.19 E | 1758 m |
| 6 | BOTTOM 2 | South facing | 15°S | 31°27 05.49 N | 069°35 48.88 E | 1576 m |

Laboratory work

Processing

After coming back from field, all fruits of one batch were washed thrice with running water and spread on clean white cloth sheet in a dark aerated room. Fruits were mixed daily and left like that for a month until hard dried fruits were obtained. Then dried fruits were grinded in an electrical grinder to obtain pasty powder.

Extraction

Powdered fruit was extracted in 80% methanol, 80% ethanol, acetone and water. Extraction was done as per method of Abideen et al. (2015). 50-gram plant material was mixed with 100 ml of each solvent and the mixture was kept in a shaking water bath (Memmert GmbH) for 3 hours maintaining the temperature at 40°C. After 3 hours the flasks were cooled to room temperature and then centrifuged at a speed of 4500 rpm for 15 minutes. Then the supernatant was collected in caped glass tubes and kept at 4°C and was used for further analysis.

Quantitative determination of Polyphenols

Total Phenolic Content (TPC)

The total phenolic content in fruit extracts were determined by applying some modifications to the procedure described by Folin and Denis (1912). Briefly, 0.5 ml of extract was placed in a test tube, distilled water was added to make the final volume of 17 ml. To the solution, 1 ml of Folin-Ciocalteu reagent was added and then after 8 minutes 2 ml of 7% sodium carbonate solution was added. After 30 minutes of incubation in the dark, the absorbance was measured at 765 nm at Shimadzu UV-Visible Spectrophotometer (UV 160). Gallic acid was used to prepare a set of standards to build a calibration curve. The total phenolic content was expressed as mg of gallic acid equivalent (GAE) per gram of dried fruit.

Total Flavonoid Content (TFC)

The total flavonoid content of fruit extracts was determined by making a few changes to the method described by Dewanto et al. (2002). Briefly, 1.0 ml of fruit extract solution was placed in a volumetric flask, 5 ml of distilled water was added to it and then 0.3 ml of 5% NaNO₂ was poured into it. The solution was mixed and incubated for 5 minutes at room temperature. Afterwards, 0.6 ml of 10% AlCl₃ was added and the second incubation was performed for 5 minutes at room temperature. Finally, 2 ml of 1M NaOH solution was added and then a volume of 10 ml was achieved by adding distilled water. The absorbance was read at 510 nm using Shimadzu UV-Visible Spectrophotometer (UV 160). All the samples were analyzed in triplicate and the results were expressed as mg/g of Catechin.

Antioxidant activity assay

The antioxidant activity of fruit extracts were measured by applying the diphenyl-picrylhydrazyl (DPPH) radical degradation method (Queiroz, 2009) with slight modifications. Briefly, 0.5 ml of fruit extract was added in an equal volume of ethanolic solution of DPPH (0.1 mM). The solution was mixed and incubated for 30 min in the dark at room temperature. All the samples were prepared and analyzed in triplicate and the absorbance was noted at 517 nm by means of a Shimadzu UV-Visible Spectrophotometer (UV 160). BHT (Butylated hydroxytoluene) was used as standard control. Inhibition of free radical by DPPH was calculated with Eq. 1.

Antioxidant activity (%) =
$$\frac{A_{blank} - A_{sampl}}{A_{blank}} x100$$
 (Eq.1)

where A_{blank} is the absorbance of the control reaction mixture without the sample and A_{sample} is the absorbance of the sample under investigation.

Statistical analysis

All the results obtained from chemical analysis were processed on MINITAB software to obtain interaction plots that could reveal any relations between the parameters under consideration *i.e.*, TPC, TFC and AA according to changing slopes, altitudes and plant development stages. STATISTIX software was used to perform analysis of variance (ANOVA) and Least Significance Difference (LSD): a statistically significant difference between groups was true if *p*-values were found less than 0.05.

Results and discussions

The fruits of *Olea ferruginea* are edible, pickled and utilized as appetizers, antidiabetics, and an olive oil source efficient in oleic acid, utilized for curing scabies, typhoid, eye burning, jaundice, biliousness, toothache and teeth caries (Zabihullah et al., 2006; Ahmad, 2007). Fruit of *Olea ferruginea* is commercially overlooked and underutilized in Pakistan due to its smaller size apparently. Free radicals impair the proper functioning of the immune system leading to the various disease conditions. Flavonoids are naturally occurring phenolic compounds in plants which have antioxidant effect (Imaga et al., 2010).

Slopes

South facing slopes were found to make TPC of olive fruits higher when compared with phenolics of olives collected from north facing slopes (*Fig. 3a,b,c*). Results showed a reverse pattern in olives extracted in 80% Methanol (*Fig. 3d*). TFC of olive fruits also show the same pattern of fluctuation as seen in TPC except for slopes *i.e.*, north facing slopes showed higher flavonoid content (*Fig. 4a,c,d*) whereas acetone extracts behaved oppositely (*Fig. 4b*).



Figure 3. Main effect plot for total phenolic contents (TPC as mg/g) of Olea ferruginea Royle fruits with respect to slope, altitude, growth stages in four extraction solvents. Slopes: 1= North facing, 2= South facing; Altitude: 1= Top 2= Middle 3= Bottom; Growth stages: 1= Green fruits; 2= Purple ripened fruits; 3= Blackish mature fruits. Solvents: a= water, b= acetone, c= 80% ethanol, d= 80% methanol

Altitude

Generally, there is an overall decline in TPC, TFC and DPPH radical scavenging activity from the highest to middle altitude while a sudden boost was noted again in fruits collected from bottom of hills *i.e.*, the lowest altitude. Pattern remained similar in all solvent extracts (*Figs. 3,4,5*).

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Figure 4. Main effect plot for total flavonoid contents (TFC as mg/g) of Olea ferruginea Royle fruits with respect to slope, altitude, growth stages in four extraction solvents. Slopes: 1= North facing, 2= South facing; Altitude: 1= Top 2= Middle 3= Bottom; Growth stages: 1= Green fruits; 2= Purple ripened fruits; 3= Blackish mature fruits. Solvents: a= water, b= acetone, c= 80% ethanol, d= 80% methanol



Figure 5. Main effect plot for Antioxidant activity (AA) as DPPH radical scavenging activity (%) of Olea ferruginea Royle fruits with respect to slope, altitude, growth stages in four extraction solvents. Slopes: 1= North facing, 2= South facing; Altitude: 1= Top 2= Middle 3= Bottom; Growth stages: 1= Green fruits; 2= Purple ripened fruits; 3= Blackish mature fruits. Solvents: a= water, b= acetone, c= 80% ethanol, d= 80% methanol

Growth stages

Total phenolics of olive fruits increase with maturation and the highest levels were observed at the last maturation stage in the present study that supports the results of Bouaziz et al. (2004) in *Olea europea*. Present study is in agreement with Sharma et al. (2017) who concluded that ripened fruits of *Olea ferruginea* can serve as a good source of natural antioxidants. Peculiarly all phytochemical attributes evaluated (TPC, TFC and AA) in present study showed a steep drop at the second maturation stage of fruit while a maximum increase was observed in the last maturation stage when fruit turned blackish from purple (*Figs. 3,4,5*) except for TPC of olive fruits in aqueous extract, TFC and AA of 80% ethanol extracts of *Olea ferruginea* that came out differently (*Figs. 3a, 4c, 5c*).

Total phenolics increase with maturation and the highest levels were observed at the last maturatiom stage in *Olea europea*. A positive corelation between total phenolic content and antioxidant activity was also recorded. Antioxidant activity also increased with fruit maturation (Bouaziz et al., 2004).

Extraction solvent

In the recovery process, the extraction solvent plays a fundamental role. In most experiments, either ethanol or methanol was used to extract plant material (Poudyal et al., 2010; Ahmad-Qasem et al., 2013; Kamran, 2016). In food products, ethanol is the predominant solvent but 80% of methanol is known to be the most active solvent in olive leaf biphenols (Malik and Bradford, 2008). Boiling water has also been used for the extraction of biphenols in some of the studies (Pereira et al., 2007; Malik and Bradford, 2008). Most of the phenolic compounds are thought to be present chiefly in free form and can be easily extracted by alcoholic solvents (Hung and Duy, 2012).

Result of analysis performed to examine level of TPC, TFC and AA of *Olea ferruginea* fruits showed significant differences when the fruit was harvested at different ripening stages and from two slopes and three altitudes (*Table 2*).

| Source | Variations | Т | otal Pher (n | nolic Co ng/g) | ntent | To | tal Flavo (n | onoid Co ng/g) | ontent | DPPH (%) | | | |
|------------------|------------|--------------------|--------------------|--------------------|--------------------|--------------------|--------------------|--------------------|--------------------|--------------------|---------------------|--------------------|--------------------|
| | | Water | Acetone | Ethanol | Methanol | Water | Acetone | Ethanol | Methanol | Water | Acetone | Ethanol | Methanol |
| Slopes | North | 12.00 ^b | 30.29 ^b | 21.00 ^b | 29.00 ^a | 10.74 ^a | 20.33 ^b | 18.89 ^a | 20.67 ^a | 11.11 ^a | 50.89 ^b | 43.00 ^b | 54.37ª |
| | South | 14.22 ^a | 33.25 ^a | 27.11 ^a | 21.56 ^b | 9.741 ^b | 28.33ª | 14.89 ^b | 19.78 ^b | 9.77 ^b | 55.22ª | 44.55 ^a | 42.44 ^b |
| | Тор | 13.67 ^b | 35.05 ^b | 26.16 ^b | 27.33 ^b | 10.61 ^b | 26.67 ^b | 17.00 ^b | 22.50 ^a | 10.67 ^b | 56.33ª | 45.33 ^b | 42.55° |
| Altitude | Mid | 10.83° | 23.83° | 17.33° | 17.94° | 6.94 ^c | 17.16 ^c | 14.00 ^c | 17.67 ^c | 7.50 ^c | 51.00 ^b | 36.33° | 47.67 ^b |
| | Bottom | 14.83 ^a | 36.44 ^a | 28.67 ^a | 30.56 ^a | 13.16 ^a | 29.16 ^a | 19.67 ^a | 20.50 ^b | 13.16 ^a | 51.83 ^b | 49.67 ^a | 55.00 ^a |
| ~ . | Green | 15.50 ^a | 29.33° | 23.33 ^b | 24.33 ^b | 10.00 ^b | 21.67 ^b | 13.83° | 20.33 ^b | 10.00 ^b | 53.67ª | 44.67 ^a | 42.83° |
| Growth stages | Purple | 12.50 ^b | 30.22 ^b | 23.00 ^b | 25.22 ^b | 8.44 ^c | 21.16 ^b | 17.67 ^b | 19.00 ^c | 8.50 ^c | 52.33 ^b | 43.67 ^b | 48.72 ^b |
| | Black | 11.33° | 35.77 ^a | 25.83ª | 26.27ª | 12.27 ^a | 30.16 ^a | 19.16 ^a | 21.33ª | 12.83 ^a | 53.16 ^{ab} | 43.00 ^b | 53.67ª |

Table 2. LSD test of TPC, TFC and AA at different slopes, altitudes, growth stages in Olea ferruginea fruits extracted in four different solvents

Different letters show significant difference between the groups (P < 0.05)

Analysis of variance study clearly shows significant differences between results of TPC, TFC and AA in response to different slope directions, altitudes and growth stages (*Tables 3,4,5*).

| Courses | DF | Water | | Acetone | | 80% Ethanol | | 80%Methanol | |
|------------------------|----|-------|------|---------|------|-------------|------|-------------|------|
| Sources | | F | Р | F | Р | F | Р | F | Р |
| Replicate | 2 | | | | | | | | |
| Slopes | 1 | 37.99 | 0.00 | 79.71 | 0.00 | 519.44 | 0.00 | 360.53 | 0.00 |
| Altitude | 2 | 43.4 | 0.00 | 578.86 | 0.00 | 657.51 | 0.00 | 372.37 | 0.00 |
| Growth | 2 | 47.39 | 0.00 | 147.66 | 0.00 | 44.47 | 0.00 | 8.22 | 0.00 |
| Slopes*Altitude | 2 | 14.34 | 0.00 | 7.71 | 0.00 | 39.32 | 0.00 | 37.45 | 0.00 |
| Slopes*Growth | 2 | 0.66 | 0.52 | 196.13 | 0.00 | 58.90 | 0.00 | 107.01 | 0.00 |
| Altitude*Growth | 4 | 2.37 | 0.07 | 12.80 | 0.00 | 1.72 | 0.17 | 3.27 | 0.02 |
| Slopes*Altitude*Growth | 4 | 3.51 | 0.02 | 83.84 | 0.00 | 2.23 | 0.09 | 14.35 | 0.00 |

Table 3. Factorial analysis of variance for total Phenolic content on the basis of different slopes, altitudes, growth stages and extraction in different solvents

Significant at p< 0.05

Table 4. Factorial analysis of variance for total Flavonoid content on the basis of different slopes, altitudes, growth stages and extraction in different solvents

| Sources | DF | Wa | nter | Acetone | | Ethanol | | Methanol | |
|------------------------|----|-------|------|---------|------|---------|------|----------|------|
| Sources | Dr | F | Р | F | Р | F | Р | F | Р |
| Replicate | 2 | | | | | | | | |
| Slopes | 1 | 5.70 | 0.02 | 366.18 | 0.00 | 269.78 | 0.00 | 32.97 | 0.00 |
| Altitude | 2 | 74.35 | 0.00 | 305.79 | 0.00 | 180.69 | 0.00 | 328.15 | 0.00 |
| Growth | 2 | 28.25 | 0.00 | 195.17 | 0.00 | 170.07 | 0.00 | 76.24 | 0.00 |
| Slopes*Altitude | 2 | 8.54 | 0.00 | 6.57 | 0.00 | 52.46 | 0.00 | 219.97 | 0.00 |
| Slopes*Growth | 2 | 40.84 | 0.00 | 51.07 | 0.00 | 247.92 | 0.00 | 496.61 | 0.00 |
| Altitude*Growth | 4 | 6.62 | 0.00 | 13.99 | 0.00 | 14.57 | 0.00 | 113.33 | 0.00 |
| Slopes*Altitude*Growth | 4 | 9.13 | 0.00 | 16.95 | 0.00 | 47.46 | 0.00 | 122.61 | 0.00 |

Significant at p< 0.05

Table 5. Factorial analysis of variance for total DPPH radical scavenging activity on the basis of different slopes, altitudes, growth stages and extraction in different solvents

| Sauraa | DF | Water | | Acetone | | Ethanol | | Methanol | |
|------------------------|----|--------|------|---------|------|---------|------|----------|------|
| Source | | F | Р | F | Р | F | Р | F | Р |
| Replicate | 2 | | | | | | | | |
| Slopes | 1 | 48.32 | 0.00 | 164.00 | 0.00 | 31.14 | 0.00 | 4033.47 | 0.00 |
| Altitude | 2 | 292.24 | 0.00 | 95.85 | 0.00 | 793.76 | 0.00 | 1479.51 | 0.00 |
| Growth | 2 | 175.48 | 0.00 | 5.28 | 0.01 | 12.07 | 0.00 | 1112.24 | 0.00 |
| Slopes*Altitude | 2 | 19.13 | 0.00 | 18.44 | 0.00 | 31.14 | 0.00 | 406.29 | 0.00 |
| Slopes*Growth | 2 | 248.62 | 0.00 | 347.73 | 0.00 | 27.33 | 0.00 | 44.62 | 0.00 |
| Altitude*Growth | 4 | 25.00 | 0.00 | 110.25 | 0.00 | 3.26 | 0.02 | 143.99 | 0.00 |
| Slopes*Altitude*Growth | 4 | 27.68 | 0.00 | 17.31 | 0.00 | 18.03 | 0.00 | 65.80 | 0.00 |

Significant at p< 0.05

Results of the present study showed that TPC of olive fruits showed the highest ranges (21-51 mg/g) when extracted in acetone *i.e.*, in opposition with the study of Kamran (2016) who stated that organic solvents are not applicable for the extraction of phenolic acids. These compounds are mainly bound by ester or glycosidic links to the cell wall (Ignat et al., 2011). Therefore, release of these compounds requires acid or base hydrolysis (Kamran, 2016).

After acetone, 80% methanol, 80% ethanol and water stood second, third and fourth for better recovery of TPC, TFC and AA as DPPH radical scavenging activity of olive fruit extracts (*Fig. 6*).



Figure 6. Main effect plot for total phenolic content, total flavonoid contents, and antioxidant activity of Olea ferruginea Royle fruits with respect to four extraction solvents. Solvents: 1= water, 2= acetone, 3= 80% ethanol, 4= 80% methanol

Conclusion

Olea ferruginea Royale is a commercially underutilized species of naturally occurring olives of Zhob. As this species of olive has small sized fruit and recently oil is main demand from the *Olea sp.* commercially it is underutilized and a way it is a neglected species of olive. But this study revealed the importance of fruit extract to be used in pharmaceutical industry as the fruit is locally used in various ailments. Present work may contribute to finding of the best harvesting time, elevation and slope direction that can help to avail maximum TPC, TFC and AA in olive fruits. It can be concluded that all three studied factors have a strong impact on the quantity of secondary metabolites and radical scavenging capacity. Safely it can be stated according to recent findings that for maximum recovery of polyphenols and a good antioxidant activity, olives can be picked when fully matured. Considering altitudinal impact, it was found that good amount of phenolics can be extracted from olives collected from both extremes *i.e.*, the highest and the lowest altitudes among sampling sites according to the present study. The study also aimed to check the potency of profitable exploitation of olives coming from an underutilized widely distributed species of District Zhob that is non-commercial right now. This can help to boost local income and benefit consumers, industries, markets, and society.

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APPENDIX

Table A1. Total Phenolic Content (mg/g) in fruits of Olea ferruginea at different altitudes, slopes, growth stages and extracted in four solvents

| | Growth | | Northe | ern slopes | | Southern slopes | | | | | |
|----------|--------------|--------------|------------|----------------------|--------------|-----------------|------------|----------------|-----------------|--|--|
| Altitude | stages | Acetone | Water | Water 80% Ethanol | | Acetone | Water | 80% Ethanol | 80% Methanol | | |
| Тор | Green fruit | 32 ± 1 | 15 ± 3 | 22 ± 3 | 25 ± 1.7 | 37 ± 1 | 17 ± 2 | 28 ± 1 | 28 ± 1 | | |
| | Purple fruit | 21 ± 1 | 11 ± 2 | 12 ± 2 | 19 ± 1 | 21 ± 1 | 15 ± 0 | 22 ± 1 | 17 ± 1 | | |
| | Black fruit | 41 ± 4 | 18 ± 2 | 24 ± 2 | 28 ± 1.7 | 24 ± 2.5 | 17 ± 0 | 32 ± 1 | 29 ± 1 | | |
| | Green fruit | 34 ± 1 | 10 ± 1.7 | 25 ± 1.7 | 32 ± 1 | 34 ± 2.6 | 15 ± 1 | 25 ± 1 | 22 ± 1 | | |
| Mid | Purple fruit | 22 ± 2.6 | 8 ± 1 | 14 ± 1.7 | 21 ± 1.7 | 22 ± 1 | 12 ± 1 | 19 ± 1 | 15 ± 0 | | |
| | Black fruit | $43\ \pm 3$ | 16 ± 1 | 27 ± 1 | 40 ± 1 | 37 ± 1 | 14 ± 0 | 28 ± 1 | 21 ± 1.5 | | |
| | Green fruit | 32 ± 2.6 | 12 ± 1 | 26 ± 1 | 32 ± 1 | 42 ± 1 | 13 ± 1 | 31 ± 2 | 25 ± 1 | | |
| Bottom | Purple fruit | 25 ± 3 | 7 ± 1 | 12 ± 2 | 22 ± 1 | 32 ± 1 | 12 ± 1 | 25 ± 1 | 17 ± 1 | | |
| | Black fruit | 33 ± 1.5 | 11 ± 1 | 27 ± 1 | 42 ± 1 | 51 ± 1 | 13 ± 1 | 34 ± 2 | 23 ± 1.7 | | |

Values presented as mean and ±Standard deviation

| | Growth | | North | ern slopes | | Southern slopes | | | | |
|----------|--------------|--------------|------------|----------------|-----------------|-----------------|-------------|----------------|-----------------|--|
| Altitude | stages | Acetone | Water | 80% Ethanol | 80% Methanol | Acetone | Water | 80% Ethanol | 80% Methanol | |
| Тор | Green fruit | 22 ± 2.6 | 9 ± 1 | 15 ± 1 | 19 ± 1 | 22 ± 1 | 11 ± 1 | $15\ \pm 1$ | 22 ± 1.0 | |
| | Purple fruit | 12 ± 1 | 5 ± 0 | 9 ± 1 | 23 ± 1 | 19 ± 1 | 9 ± 1 | 11 ± 1 | 13 ± 0 | |
| | Black fruit | 27 ± 2.6 | 11 ± 1 | 12 ± 1 | 25 ± 0 | 28 ± 7 | 15 ± 0 | 21 ± 1 | 20 ± 1.0 | |
| | Green fruit | 18 ± 2 | 11 ± 0 | 19 ± 1 | 27 ± 1 | 32 ± 1 | 9 ± 1.1 | 18 ± 2.6 | 21 ± 1.7 | |
| Mid | Purple fruit | 11 ± 1 | 6 ± 1 | 21 ± 1 | 22 ± 1 | 19 ± 1 | 6 ± 1 | 8 ± 0 | 11 ± 1.0 | |
| | Black fruit | 22 ± 3 | 9 ± 1 | 25 ± 0 | 15 ± 1 | 25 ± 2 | 10 ± 1 | 15 ± 1 | 18 ± 1.7 | |
| | Green fruit | 25 ± 2.6 | 15 ± 1 | 19 ± 1 | 21 ± 0 | 41 ± 2.6 | 9 ± 1 | 16 ± 1 | 25 ± 1.7 | |
| Bottom | Purple fruit | 17 ± 1 | 11 ± 0 | 23 ± 2 | 16 ± 1 | 25 ± 1 | 8 ± 0 | 12 ± 1 | 21 ± 1.0 | |
| | Black fruit | 29 ± 1 | 23 ± 1 | 27 ± 1 | 18 ± 1 | 44 ± 2 | 11 ± 2 | 18 ± 1.7 | 27 ± 1.0 | |

Table A2. Total Flavonoid Content (mg/g) in fruits of Olea ferruginea at different altitudes, slopes, growth stages and extracted in four solvents

Values presented as mean and ±Standard deviation

| | Growth | | Northe | ern slopes | | Southern slopes | | | | |
|-----------|--------------|--------------|------------|----------------|-----------------|-----------------|--------------|----------------|-----------------|--|
| Altitudes | stages | Acetone | Water | 80% Ethanol | 80% Methanol | Acetone | Water | 80% Ethanol | 80% Methanol | |
| Тор | Green fruit | 58 ± 2.6 | 35 ± 0 | 44 ± 1 | 26 ± 2.6 | 66 ± 2 | 42 ± 1 | 47 ± 1 | 36 ± 1.0 | |
| | Purple fruit | 40 ± 1 | 48 ± 1 | 35 ± 2 | 45 ± 1 | 43 ± 2 | 32 ± 1 | 40 ± 1 | 23 ± 0 | |
| | Black fruit | 52 ± 1 | 62 ± 1 | 50 ± 1 | 60 ± 1 | 53 ± 2.6 | 48 ± 1 | 52 ± 2 | 42 ± 1.0 | |
| | Green fruit | 60 ± 1 | 35 ± 0 | 42 ± 2 | 40 ± 2 | 52 ± 2 | 30 ± 0 | 48 ± 1 | 39 ± 1.0 | |
| Mid | Purple fruit | 54 ± 2.6 | 50 ± 1 | 38 ± 2 | 59 ± 2 | 32 ± 1 | 42 ± 0 | 35 ± 0 | 22 ± 1.0 | |
| | Black fruit | 52 ± 3.6 | 64 ± 0 | 47 ± 1 | 65 ± 1 | 44 ± 0 | 58 ± 1 | 52 ± 1 | 50 ± 0 | |
| | Green fruit | 57 ± 2.6 | 33 ± 2 | 44 ± 1.7 | 56 ± 2 | 45 ± 1 | $21 \ \pm 1$ | 47 ± 1 | 46 ± 1.7 | |
| Bottom | Purple fruit | 51 ± 2 | 52 ± 0 | 38 ± 1 | 61 ± 1 | 32 ± 0 | 19 ± 1 | 32 ± 1 | 23 ± 1.0 | |
| | Black fruit | 62 ± 2 | 63 ± 1 | 49 ± 1 | 65 ± 1 | 48 ± 1 | 25 ± 0 | 48 ± 1 | 48 ± 1.0 | |

Table A3. Antioxidant activity as DPPH radical scavenging activity (%) in fruits of Olea ferruginea at different altitudes, slopes, growth stages and extracted in four solvents

Values presented as mean and ±Standard deviation